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LESSON 1 SOME BASIC CONCEPT OF CHEMISTRY
Chemistry[Type Text] [Type Text] [Type Text] LESSON

1 SOME BASIC CONCEPT OF CHEMISTRY Chemistry 1)
40 ml of Methane is completely burnt using 80 ml of
Oxygen at room temperature. The volume of gas left
after cooling to room temperature is (a) 40 ml CO₂
Gas (b) 40 ml CO₂ Gas and 80 ml H₂O Gas (c) 60 ml
CO₂ Gas and 60 ml H₂O Gas (d) 120 ml CO₂ Gas Mar
2th, 2024

UNIT 1 SOME BASIC CONCEPTS OF

CHEMISTRY Chemistry 2 Prefixes in SI System Multiple
Prefix Symbol 10⁻¹² Pico P 10⁻⁹ Nano N 10⁻⁶ Micro μ
10⁻³ Milli M 10⁻² Centi C 10⁻¹ Deci D 10 Deca Da 10²
Hecto H 10³ Kilo K 10⁶ Mega M 10⁹ Giga G 10¹² Tera
T Mass and Weight-- Mass of a substance is the

amount of matter present in it while weight is the
force exerted by gravity on an object. The mass of a
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Organic Chemistry, with an emphasis of topics of
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Chemistry In which any number can be represented in
the form $N \times 10^N$ (where N is an exponent having
positive or negative values and N can vary between 1
to 10). E.g. We can write 232.508 as 2.32508×10^2 .
In scientific notation. Similarly, 0.00016 can be

Written As 1.6×10^{-4} . Precision Refers Jun 3th, 2024
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Balancing Chemical Equations. The Following Four Steps Can Be Used As A Guide To Balance Chemical Equations. Example: Sulfuric Acid Reacts With Sodium Hydroxide To Create Sodium Sulfate And Water. Balance This Chemical Reaction. Step 1: Write An Unbalanced Equation, U Feb 2th, 2024
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