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Every Second. The Acceleration Is A Constant Value Of 10 M/s/s; Thus, Choice B
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$\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$ Actual yield = 1.45g Theoretical yield = 5.95g $\text{KClO}_4 \times 1 \text{ mol } 122.55 \text{ g} \times$
 $2 \text{ O}_2 \times 32.00 \text{ g } 1 \text{ mol } = 3.11 \text{ g O}_2$ %Yield = $1.45 \text{ g } 3.11 \text{ g} \times 100\% = 46.6\%$ 34.
 $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$ Percent purity = 62.8%? g pure FeCO_3 4th,

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