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 Chemistry 21.1 Radioactivity •When Nuclei Change Spontaneously, Emitting  
 Energy, They Are Said To Be Radioactive. •Nuclear Chemistry Is The Study Of  
 Nuclear Reactions And Their Uses. •Nucleons Are Particles In The Nucleus: • $p^{+}$ :  
 Proton • $n^0$ : Neutron •Atomic Number Is The Number Of  $P^{+}$ . Mar 15th, 2024.  
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 Nuclei Can Either Decay Into A Stable Nucleus Of Different  $N$  And  $Z$ , Or Can Return  
 To The Ground State From An Excited State. Three Different Types Of Decay  
 Particles: 1. Alpha Particles, Which Consist Of Two Neutrons And Two Protons, And  
 Are Nuclei Of 2. Electrons Mar 27th, 2024 Chapter 22 Review Nuclear Chemistry  
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1/11/2015 Worksheet #1: Radioactivity Chemical Reactions Involve Changing One  
Substance Into Another Substance By Rearranging Atoms. However, During A  
Chemical Reaction Atoms Of One Element Cannot Change Into Atoms Of Another  
Element. The May 4th, 2024Practice(Packet:(Nuclear(Chemistry( Regents  
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Unknown Particle 3. Use The Law Of Conservation Of Charge To Obtain The Charge  
Of The Unknown Particle 4. If The Unknown Particle Is A Form Of Radiation, find Its  
Symbol On Table O 5. If The Unknown Particle Is An Element, find Its May 28th,  
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