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Bonding: Covalent And Ionic (metallic Bonding Discussed Slightly In Ch 11)  
Examples: Cl<sub>2</sub> HCl NaCl Bond Classification Scheme Pure Covalent: No Difference In Electronegativity. Polar Covalent: Moderate Difference. Ionic: Large Difference. Really A Continuum (Fig. 8.7) Electronegati 3th, 2024

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### **IONIC BONDING Name Ionic Bonding Occurs When A Metal ...**

IONIC BONDING Name Ionic Bonding Occurs When A Metal Transfers One Or More Electrons To A Nonmetal In An Effort To Attain A Stable Octet Of Electrons. For Example, The Transfer Of An Electron From Sodium To Chlorine Can Be Depicted By A Lewis Dot Diagram, Calcium Would Need Two Chlorine Atoms To Get Rid Of Its Two Valence Electrons, 1th, 2024

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Objectives For Today: Explain Properties Of Ionic And Metallic Compounds Using

Diagrams. Bellwork: Get A White Board (one Per Table)and Answer The Questions Below: Draw The Lewis Dot Structure Of The Formation Of A Compound Made A Calcium And Phosphorus. What Is The Charge Of Each Ion Formed? W 2th, 2024

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Core Electrons -those In The Energy Levels Below. 7 Keeping Track Of Electrons ...  
-Have The Same Outer Electron Configuration.-Have The Same Valence Electrons.  
ElectronsThe Number Of Valence Electrons Are Easily Determined: The Group  
Number For A Representative Element ... -Mg, Zn, Cd. 5 3th, 2024

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## **Introduction To Ionic Bonds I. Ionic Bonds III. Metallic ...**

Metallic Bonding Is Found In Elemental Metals And In Mixtures Of Metals Called Alloys. 1. What Is A Metallic Bond? Explain How The Ions And Electrons Are Arranged. 2. List Some Properties Of Metallic Bonds. 3. What Is An Alloy? 4. Identify The Following Compounds As Metallic, Ionic Or Covalent: A. NaCl B. Cl<sub>2</sub> C. Au D. [BrO<sub>3</sub>]<sup>-1</sup> E. Mg<sub>3</sub>N<sub>2</sub> F ... 3th, 2024

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