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Theoretical yield = 5.95g KClO₄ × 1mol / 122.55g × 2O₂ / 1KClO₄ × 32.00g / 1mol = 3.11g O₂ %Yield = 1.45g / 3.11g × 100% = 46.6% 34.
4FeCO₃ + O₂ → 2Fe₂O₃ + 4CO₂ Percent purity = 62.8%? g pure FeCO₃ 1th, 2024

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