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 Investigation Is To Experimentally Determine (1) PK_a (and Thus K_a) Of The Acid In A
 Buffer And Thus The Buffer Range, (2) Investigate The Buffer Capacity Of 4th, 2024.
 Section 19.1. Acid-Base Buffer SolutionsIf We Add CH_3COO^- -SHIFT Net Effect: $[\text{H}_3\text{O}^+]$
 Decreases PH Increases Also, $[\text{OH}^-]$ Increases POH Decreases I.e.,a Weak Acid
 And Its Conjugate Base Or A Weak Base And Its Conjugate Acid. 19-2 Example:
 Addition Of NH_4Cl To NH_3 2th, 2024A Simple And Universal Method For Making Up
 Buffer SolutionsTBoyer, R F (1986) 'Modern Experimental Biochemistry', Addison-
 Wesley Publishing Co, Reading, MA 2Bohinski, R C (1983) 'Modern Concepts In
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