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During A Chemical Reaction. (Clarification Statement: Limited To Chemical Equations That Include Binary Ionic And Covalent Compounds And Will Not Include Equations Containing Polyatomic Ions.) 1 2 3 Balancing Equations "Bubble Method" Follow The Steps In The Pictures - Labeled 1,2,3 To Balance 4th, 2024

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Balancing Chemical Equations Process For Balancing Chemical Equations: 1. Determine Correct Chemical Formulas Of All Reactants And Products 2. Start With "heavier" Atoms—balance Number Of These On Reactant And Product Sides Of Equation 3. If Elements Appear In Equati 3th, 2024

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11. Zinc Reacts With Silver Nitrate To Produce Zinc Nitrate And Silver. (a) Zn + 2 AgNO 3 → Zn (NO 3)2 + 2 Ag (b) Single Replacement 12. Glucose (C 6H12 O6) Reacts With Oxygen Gas To Produce Carbon Dioxide And Water. (a) C6H12 O6 + 6 O2 → 6 CO 2 + 6 H2O (b) Combustion 13. Potassium Oxide Reacts With Water To Produce Potassium Hydroxide. 2th, 2024

#### 8 Chemical Equations And Reactions

Chemical Equations And Reactions MIXED REVIEW SHORT ANSWER Answer The Following Questions In The Space Provided. 1. B A Balanced Chemical Equation Represents All The Following Except (a) Experimentally Established Facts. (b) The Mechanism By Which Reactants Combine To Form Products. (c) Identities Of Reactants And Products In A Chemical Reaction. 2th, 2024

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