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Ftisfertf: Form WS5.13 . 1A Percent Yield CHEMICAL FORMULAS AND EQUATIONS Page 2 Calculate The Percent Yield In Each Of The Problems Below. (NOTE: Equations May Not Be Balanced.) 1. 3.25×10^3 Kg Of Iron III Oxide Are Treated In A Blast Furnace With 1.50×10^3 Kg Of Carbon Monoxide To Form 1.30×10^3 kg Of Pure Iron. $[\text{Fe}_2\text{O}_3 + \text{CO} \rightarrow \text{Fe} + \text{CO} \dots$ 2th, 2024

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With Chemical Formulas And Equations 2 | 1 . 2 ...

Percentage Composition Example: A Sample Of A

Liquid With A Mass Of 8.657 G Was ... •Smallest Is

3.722×10^{-3} Mole •C = •H = Step 3: Divide All

Number Of Moles By The Smallest 2th, 2024

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