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5. List The Seven Diatomic Molecules. Chapter 9:
Chemical Names & Formulas 1. Determine Whether
The Following Compounds Are Ionic Or Covalent, And
Then Name Them. CaSO_4 , N_2O_4 , NH_4NO_3 , CaS , Li_3PO_4 , NH_3 , $\text{Cu}(\text{OH})_2$ 2. 2. Complete The Following
Table And Name Each Ionic Compound Formed. + Jan
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C. Ag-107, Au-197 And Ar-40 11. List The Order That You Fill The Energy Levels And Orbitals For Electron Configuration. 12. Fill In The Orbital Diagram Below For Tungsten And Then Write The Full Electron Configuration Below. W: 13. Which Of The Following Is An Incorrect Orbital Diagram? Why? Jan 1th, 2024

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18. Write The Abbreviated (noble Gas) Electron Configuration For The Following Atoms: A. Silver B. Copper C. Tungsten D. Titanium 19. Define Ground State. 20. Define Excited State. 21. How Do Atoms Jump From Ground State To Excited State And Back? (Remember Our Flame Test L May 1th, 2024

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2. How Does Calcium Obey The Octent Rule When Reacting To Form Compounds> A. It Gains Electrons B. It Gives Up Electrons C. It Does Not Change Its Number Of Electrons D. Calcium Does Not Obey The Octet Rule 3. What Is The Formula When Potassium And Sulfur Combine? K. 2S 4. What Is The F Mar 1th, 2024

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9. Find A Unit Vector In R^2 That Is Orthogonal To $\begin{pmatrix} 1 \\ 2 \end{pmatrix}$.

Solution: We Want To Find A Vector $\sim v = v_1 v_2$ Such That $\sim v \cdot 1 2 = 0$. Evaluating This Dot Product Gives The Equation $v_1 + 2v_2 = 0$ So $v_1 = -2v_2$: Thus, Any Vector Of The Form $\sim u = v_2 -2 1$ Is Orthogonal To $1 2$.
Let Us Take May 1th, 2024

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