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The Type. $A + B \rightleftharpoons C + D$. $K_c = \frac{[C][D]}{[A][B]}$ Le Chatelier's Principle, Catalysts Changes In Equilibria.
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Pre-Lab (Purpose, Pre-Lab Questions, Hypothesis, Data Table Set-up) Is Due The Day
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2 Hr And 20 Min = 4 Half-lives Half-life Time Elapsed Mass Left 1st 35 Min 2.50 G
2nd 70 1.25 G 3rd 105 0.625 G 4th 140 0.313 G Rate = $k[\text{sugar}]$ And $k = 3.3 \times 10^{-4}$
Sec Sec^{-1} . Half-life. Half-life Is 35 Min. Start With 5.00 G 4th, 2024

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Reactants And Characteristics Of The Reaction Mechanism. As A Result, The ASM Has Been Successfully Applied By Various Research Groups, On A Wide Range Of Chemical Reactions, Such As Nucleophilic Substitution,⁶ Cycloadditions,⁷ Oxidative Addition,⁸ And Many Other Processes In Organic 2th, 2024

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