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(b) He_2 Is Not Stable But He^+_2 Is Expected To Exist. (c) Bond Strength Of N_2 Is Maximum Amongst The Homonuclear Diatomic Molecules Belonging To The Second Period. Thus, Dinitrogen (N_2) Molecule Contains Triple Bond And No Other Molecule Of Second Period Have More Than Double Bond. Hence, Bond Strength Of N_2 Is Maximum Amongst The Homonuclear May 12th, 2024

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