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33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$ Actual yield = 1.45g Theoretical yield = 5.95g $\text{KClO}_4 \times 1 \text{ mol} = 122.55 \text{ g} \times 2 \text{ O}_2 = 32.00 \text{ g} \times 1 \text{ mol} = 3.11 \text{ g}$
 $\% \text{ Yield} = \frac{1.45 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$
34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$ Percent purity = 62.8%
pure FeCO_3 3th, 2024Chemistry 12, Stoichiometry Review Problems, R. Janssen ...In A Camping Stove? 2. How Many Moles Of O_2 Molecules Should Be Supplied To Burn 1 Mol Of CH_4 ... Was At One Time Considered For Use As A Rocket Fuel. ... How Many Formula Units Of Calcium Oxide (CaO) 14th, 2024Chapter 9 Review Stoichiometry Section 1 Answer KeyChapters 2 And 3): A) Element B) ...Chapter 3 Chemical Reactions And Reaction Stoichiometry: Chapter 4 Aqueous Reactions And Solution Stoichiometry: Chapter 5 Thermochemistry: Chapter 6 Electronic Structure Of Atoms: Chapter 7 Periodic Properties Of The Elements: Chapter 8 Basic Concepts Of Chemical Bonding: Chapter 9 Molecular Geometries And ... 19th, 2024.

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