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In A Limited Supply Of Pure Oxygen. 0.50g Of The
Carbon Combusts To Form CO₂ And 0.50 G Of The
Carbon Combusts To Form CO. The Resultant Mixture
Of CO₂ And CO Is Passed Through Excess Na₄th,
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to-one Molar Relationship, So 0.0322 Mol Of Aniline
Produces 0.0322 Mol Of Acetanilide. The Amount Of
Acetic Anhydride Does Not Play A Role, Since It Is In
Excess. Then, N (N) 0.0322 .aniline N Aceta II. Ide Mol
We Find The Molar Mass Of Acetanilide: (N) () 12 8 1
9 14 1 16 1 135(). 89 M Aceta ... 2th, 2024Unit 13:
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