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Stoichiometry Volume Problems Worksheet Answers Stoichiometry: Mixed Problems (KEY) 1)  $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$  What Volume Of  $\text{NH}_3$  At STP Is Produced If 25.0 Of  $\text{N}_2$  Is Reacted With An Excess Of  $\text{H}_2$ ? 3 3 3 2 3 2 2 2 40.0L  $\text{NH}_3$  1mol  $\text{NH}_3$  22.4 Jun 3th, 2024

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A Practice Problem On Stoichiometry -- ANSWERS Consider The Reaction Represented By The Equation Below For All Parts Of This Question:  $2 \text{BrCl}_3 \rightarrow 3 \text{Cl}_2 + \text{Br}_2$  (a) If 4 Moles Of  $\text{BrCl}_3$  Reacts According To The Equation, How Many Moles Of  $\text{Cl}_2$  Will Be Formed? How Many Moles Of  $\text{Br}_2$  Will Be Formed? Answers: 6 Mol  $\text{Cl}_2$  And 2 Mol  $\text{Br}_2$  Setups/Work: Jul 3th, 2024

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