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PH And POH - Ms. Mogck's Classroom - Ms. Mogck's Classroom

KEY Chemistry: PH And POH Calculations Part 1: Fill In The Missing Information In The Table Below.

PH [H 3 O 1+]	POH [OH 1-]	ACID Or BASE?	3.78	1.66	$\times 10^{-4}$	M	10.22	6.03	$\times 10^{-11}$																										
M Acid	3.41	3.89	$\times 10^{-4}$	M	10.59	2.57	$\times 10^{-11}$	M Acid	8.81	1.55	$\times 10^{-9}$	M	5.19	6.46	$\times 10^{-6}$	M	Base	8.69	2.04	$\times 10^{-9}$	M	5.31	4.88	$\times 10^{-6}$	M	Base	8.46	3.47	$\times 10^{-9}$	M	5.54	2.88	$\times 10^{-6}$	M	Base

1th, 2024

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