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Tert-butoxide Ion Is Very Bulky, Or Sterically Hindered,
And Is Less Capable Of Interacting With The Solvent.
The Ethoxide Ion Is Not As Sterically Hindered So It Can
Accommodate More Solvent Interactions. As A Result
Ethoxide Is Better Solvated And Is Therefore More
Stable Than Tert-butoxide (Figure 3.7). Apr 24th,

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