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Colligative Properties- Page 1 Lecture 4: Colligative ...

ΔT_b Raises Boiling Point Of Water From 100 °C To 100.64 °C • ΔT_f Lowers Freezing Point Of Water From 0 °C To -2.32 °C • π Raises Osmotic Pressure From 1 atm To 28.6 atm While The Temperature Was Changing By 1 °C 3th, 2024

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Ions In Aqueous Solutions And Colligative Properties MIXED REVIEW SHORT ANSWER Answer The Following Questions In The Space Provided. 1. Match The Four Compounds On The Right To Their Descriptions On The Left. B An Ionic Compound That Is Quite Soluble In Water (a) HCl C An Ionic Compound That Is Not Very Soluble In Water (b) NaNO₃ 2th, 2024

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12.3 Colligative Properties - REMONDINI

Colligative Properties For Molecular Solute Vs. Ionic Solute. Example: 1.0 M Of Sugar & 1.0 M Of Salt Show Different ΔT_b And ΔT_f Even Though The Molality

Concentrations Of The Solute Are Identical. Note That; $\Delta T = M \cdot K_{f,b}$ Where K Is Only A Function Of The Solvent !!! Reason For The Difference In Colligative Properties Must Be Based On ... 2th, 2024

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