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3)  $2(aq) Y$  C.  $K(s) H_2SO_4(aq) Y$  D.  $Zn(s) NaOH(aq) Y$  SECTION 11.3 REACTIONS IN AQUEOUS SOLUTION 1. Write The Net Ionic Equation For The Reaction Between

Aqueous Barium Nitrate,  $\text{Ba}(\text{NO}_3)_2$ , And Sodium Sulfate,  $\text{Na}_2\text{SO}_4$ . 2. Magnesium Reacts With HCl To Form Hydrogen And Magnesium Chloride. Write The Balanced Net Ionic Equation For This Reaction. 3. 3th, 2024

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A Chemical Reaction Can Be Concisely Represented By A Chemical 1.. The Substances That Undergo A Chemical Change Are The 2.. The New Substances Formed In A Chemical Reaction Are The 3.. In Accordance With The Law Of Conservation Of , 4. A Chemical Equation Must Be Balanced. When Balancing An 5. Equation, You Place In Front Of Reactants And ... 2th, 2024

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(b) This Is A Chemical Reaction As A Gas Is Created (carbon Dioxide) During The Reaction. (a) + H<sub>2</sub>O (b) The Bottle Could Be Heated Up Or Opened To Reverse The Reaction. CHECK YOUR LEARNING Suggested Answers 1. The Arrow Shows Which Way The Chemical Reaction Is Going. 2. 2th, 2024

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Have To Move Closer To One Another Than They Would If They Were Sharing One Or  
Two Pairs Of Electrons. Triple Bon 2th, 2024

### **11.1 Describing Chemical Reactions 11**

Figure 11.1 The Separate Ingredi-ents Of The Bread Are Now Combined Into One  
Product~the Loaf Of Bread. Connecting To Your World Section 11.1 Describing  
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