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Electrochemistry 1. Consider The Galvanic Reaction

Between The Two Half Cells,  $\text{Pb}|\text{Pb}^{2+}(\text{aq})$  And  $\text{Ag}|\text{Ag}^{+}(\text{aq})$ . 1(a) (6 Pts) Determine The Spontaneous Reaction. Write A Balanced Overall Equation And Determine  $E^{\circ}$ . 1(b) (4 Pts) Label The Electrodes And Solutions With The Following Labels: "Pb," " $\text{Pb}^{2+}$ ," "Ag," And " $\text{Ag}^{+}$ ." Make 3th, 2024.

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