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1 PV=RTtffiD 1.013 X Pa 8.31 X 103 X 298/(1.013 X 105 24.4 L (1 1 E, 250D7XDEEI*fi1.00 G/ 5, 1 Mol P = RTW = 8.31 X 103 X 298/(18/1000) = 1.38 X 10B Pa $-\tilde{n}$, Mol/L C 1 Mol 1 = 55.6 1/(1 + 1 Mol 18 Mar 15th, 2024

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