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Reactions 1.1 Rate Of Reaction And Dependence On  
Temperature We Will Once Again Look At The  
Formation Of Ammonia ( $\text{NH}_3$ ) From Nitrogen And  
Hydrogen (see Section Chemical Equilibrium Of The  
Thermodynamics Chapter). This Reaction Follows The  
Equation:  $\text{N}_2 + 3\text{H}_2 \rightleftharpoons 2\text{NH}_3$  (1)  $H_0 = 92 \text{ KJ Mol}$   $S_0 =$   
 $192 \text{ J Mol K}$  May 1th, 2024 Chemical Engineering,  
CM3510: Chemical Reaction ...Grading: Exam 1 15%  
Exam 2 15% Exam 3 15% Final Exam 20% Research  
Paper 10% Homework 20% Class Participation 5% Last  
Year, The Following Grading Scale Was Used:  $A \geq 92$   
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1  $A \rightarrow B \rightarrow C$  2.  $A \rightarrow B \rightarrow C$  3.  $A \rightarrow B \rightarrow C$  4.  $A \rightarrow B \rightarrow C$

1. Use Molar Flow Rates And Concentrations; DO NOT Use ... Jul 2th, 2024.

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