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Answers: Moles And Stoichiometry Practice Problems Answers: Moles And Stoichiometry Practice Problems 1) How Many Moles Of Sodium Atoms Correspond To  $1.56 \times 10^{21}$  Atoms Of Sodium?  $1.56 \times 10^{21}$  Atoms Na  $\times 1$  Mol Na =  $2.59 \times 10^3$  Mol Na  $236.022 \times 10$  Atoms Na 2) Determine The Mass In Grams Of Each Of The Following: A. 1.35 Mol Of Fe  $1.35$  Mol Fe  $\times 55.845$  G Fe =  $75.4$  G Fe 1 Mol Fe B. 24.5 Mol O May 2th, 2024 Stoichiometry Practice Problems Pdf Answers Stoichiometry Practice Problems Pdf Answers Balance The Following Chemical Reactions: A.  $2 \text{CO} + \text{O}_2 \rightarrow 2 \text{CO}_2$  B.  $2 \text{KNO}_3 \rightarrow 2 \text{KNO}_2 + \text{O}_2$  C.  $2 \text{O}_3 \rightarrow 3 \text{O}_2$  D.  $\text{NH}_4\text{NO}_3 \rightarrow \text{N}_2\text{O} + \dots$  Jan 2th, 2024 A Practice Problem On Stoichiometry -- ANSWERS A Practice Problem On Stoichiometry -- ANSWERS Consider The Reaction Represented By The Equation Below For All Parts Of This Question:  $2 \text{BrCl}_3 \rightarrow 3 \text{Cl}_2 + \text{Br}_2$  (a) If 4 Moles Of  $\text{BrCl}_3$  Reacts According To The Equation, How Many Moles Of  $\text{Cl}_2$  Will Be Formed? How Many Moles Of  $\text{Br}_2$  Will Be Formed? Answers: 6 Mol  $\text{Cl}_2$  And 2 Mol  $\text{Br}_2$  Setups/Work: Feb 1th, 2024.

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