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**% Yield = Actual Yield X 100 % Theoretical Yield  
Moles Of ...**

$\Delta G^\circ = -RT \ln K$  Standard Emf Of An Electrochemical  
Cell:  $E^\circ_{\text{Cell}} = E^\circ_{\text{Ox}} - E^\circ_{\text{Red}} = E^\circ_{\text{Cathode}} - E^\circ_{\text{Anode}}$   
Standard Free Energy Change:  $\Delta G^\circ = -nFE^\circ_{\text{Cell}}$  Where  
F Is The Faraday Constant Relationship Of The  
Standard Emf Of The Cell To The Equilibrium Constant:  
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Mean Values In Column Not Having The Same Letter  
Vary Significantly At  $P = 0.05$ . Humidity Is Low In  
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Mostly Range Deviate Very Little From Normal Range  
In Cropping Season. An Interactive Effect Between  
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