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1. Examples of Free Energy Calculations Free Energy Calculations Are Carried Out In Much The Same As Enthalpy calculations. There Are Tables Of Standard Free Energies Of Formation Of Compounds.Elements In Their Standard States Are assigned Zero as their Gibbs Free Energy Of Formation. Consider The Following Chemical Reaction: H₂(g) + Cl₂(g) → 2HCl(g). 2th, 2024

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Example 9-4. Absorption Of SO₂ The Molar Density Of The Feed Gas May Be Determined From The Formula $(\text{mol/L}) = 12.2P(\text{atm})/T-1(\text{K})$ Where P And T Are The Absolute Pressure And Temperature Of The Gas. The Following Data Are Taken For The Feed Gas: T=750F, P=150 Psig, H=210 Mm, R=82.4 And For The Outlet Gas, R=11.6. Determine The (a) Mass Fraction Of SO₂ 2th, 2024

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Ash Drum, The Size Of The Drum And The Liquid And Vapor Compositions And Flow Rates. Which Of These Variables Are Specified And Which Need To Be Chosen Depend On The Application. Normally, The Feed Is Specified, i.e. The Flow Rate F, Composition Z, Pressure P F And Enthalpy H F Are Known 1th, 2024

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