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(2) To Determine The Proper Number Of Sig. Figs When Multiplying Or Dividing The Measurement With The Least Number Of Sig. Figs Is Boss G. If All This Is Confusing, Use The (A)tlantic - (P)acific Rule. If The Decimal Point Is (A)bsent - Start Counting Significant Figures From The Atlantic 2th, 2024

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2. Element J Ha	as 27 Protons And 32 Neutrons. Element L Has 27 Protons And 33 Neutrons	_ 3. Element X Has	17 Protons And 18 Neutrons. Element Y Has 18 Protons And 17
Neutrons	_ 4. Element Q Has 56 Protons And 81 Neutrons. Element F Has 56 Protons And 8	2 Neutrons 5	5. Element T Has An Atomic Number 20 And Mass Number 40.
Element Z 4	th. 2024		

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D. Uranium-235: 92 Protons, 143 Neutrons E. Carbon-12: 6 Protons, 6 Neutrons 3. Most Of An Atom's Mass Is Concentrated In The Nucleus. The Number Of Electrons And Protons Is The Same But Electrons Are So Light They Contribute Very Little Mass. The Mass Of The Proton Is 1,835 Times The Mass Of The Electron. Neutrons Have A Bit More Mass Than ... 4th, 2024

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Oct 22, 2014 · B. Do All Isotopes Of An Element Have The Same Mass Number? Give At Least 1 Example Or Counter-example From Model 1 That Supports Your Answer. NO, CARBON-12 AND CARBON-13 ARE ISOTOPES OF CARBON BUT HAVE DIFFERENT MASS NUMBERS. 12. Considering Your Answers To Question 11, Work With Your Group To Write A Definition Of Isotope Using A Sentence. 4th, 2024

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