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Gases KCl And Oxygen ( $\text{O}_2$ .)  $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$  C.  
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Chemistry: Stoichiometria - Tab 1 Indications: Solve  
Each Of The Following Problems. You Stayed On The  
Right Site To Start Getting This Information. In The

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Mixed Stoichiometry Problems 1. How Many Moles Of  $H_2$  Would Be Required To Completely React With  $O_2$  To Produce 5 Moles Of Water? 5 Mol  $H_2$  2.  $H_2SO_4 + NaOH \rightarrow Na_2SO_4 + H_2O$  A. Balance This Equation B. What Mass Of  $H_2SO_4$  Would Be Required To React With 0.75 Mol Of  $NaOH$ ? 37g 3. What Mass Of  $NO_2$  Is Formed When  $NO$  Reacts With 384 G Of  $O_2$ ? Jan 5th, 2024

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Note Questions Of The Author: Practical Of Mixed Stoichiometry Since Everyone Is Doing A Fabulous Job On Tasks. I Thought It Would Be Fun To Review The Balancing Chemical Equations. For These Problems, Balance The Equation First Then Do Stoichiometry As Described In The Question. 7:30 PM - A ... Jan 5th, 2024

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