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[Feb 14th, 2024 Study Of Reaction Rates: Clock Reaction Lab Chemical Reaction- Called A Clock Reaction- Will Be Used To Determine Quantitatively The Influence Of Concentration On Rate. The First Reaction Used Is The Oxidation Of Iodide Ions By Hydrogen Peroxide In Aqueous Solutions: $2\text{I}^-(\text{aq}) + \text{H}_2\text{O}_2(\text{aq}) + 2\text{H}^+(\text{aq}) \rightarrow \text{I}_2(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$ May 3th, 2024.

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