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Sheet 1 Directions: Solve Each Of The Following Problems. Show Your Work, Including Proper Units, To Earn Full Credit. 1. Silver And Nitric Acid React According To The Following Balanced Equation:  $3 \text{Ag}(s) + 4 \text{HNO}_3(aq) \rightarrow 3 \text{AgNO}_3(aq) + 2 \text{H}_2\text{O}(l) + \text{NO}$  ... Feb 1th, 2024.

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Chemistry 11 Stoichiometry Review Worksheet Name: \_\_\_KEY\_\_\_  
33.  $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$   
Actual yield = 1.45g Theoretical yield = 5.95g  
 $\text{KClO}_4 \times 1 \text{ mol } 122.55 \text{ g} \times 2 \text{ O}_2 \times 1 \text{ mol } 32.00 \text{ g}$   
 $1 \text{ mol } = 3.11 \text{ g O}_2$  %Yield =  $\frac{1.45 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$   
34.  $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$   
Percent purity = 62.8%  
pure  $\text{FeCO}_3$  Jul 2th, 2024  
Practice Work 53 Stoichiometry-04 Mixed Stoichiometry ...  
Practice Work 53 - Stoichiometry-04 Mixed Stoichiometry Problems  
General Information You Will Need A Periodic Table, Your Stoichiometry Notes, And Appendix 12 For This Assignment.  
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