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Suggested Stage 2 Chemistry 2013 SACE Board Of SA Exam Solutions Question 1 Possible Solution Marks Comments (a) (i) Electrolytic 1 (ii) From Left To Right In External Circuit 1 (iii) Left Electrode 1 (b) (i) D Block 1 Note: Must Be Lower Case Letter 2(ii) (1) $\text{Cr} + 2\text{O} \rightarrow \text{Cr}_2\text{O}_3$ - = +6 Cr = 0 2 (2) Reduction 1 Cr(iii) + 6HCl 2 O 3th, 2024

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A Reaction Mixture Is Prepared Containing 0.60 Mol Of Aluminum And 1.20 Mol Of Manganese Dioxide. The Mixture Is Heated Until One Of The Reactants Has Been Completely Consumed According To The Equation: $2\text{Al} + 3\text{MnO}_2 \rightarrow 3\text{Mn} + \text{Al}_2\text{O}_3$ What Quantity Of Which Reactant Remains Uncombined? (A) 0. 1th, 2024

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Manual Transmission Not Available 5-speed Automatic Transmission CVT 6-speed Fuel Capacity (gallons) 17.2 18.5 Required Fuel Regular Unleaded Regular Unleaded EPA Fuel Economy, City/Highway/Combined (mpg W/automatic Transmission**) 27/36/30 22/31/25 ECON Mode Standard Not Available 2th, 2024

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Wait For The Waiter To Bring A Requested Item (plate Or Bowl) To The Table When It Is Required. Assume Th 1th, 2024

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