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And Contains 11.1 G Of Pure Silver. What Is The

Percentage Weight By Weight Concentration Of Silver
In The Metal? Solution $M_{Ag} = 11.1 \text{ G}$ $M_{Alloy} = 12.0 \text{ G}$
 $C_{Ag} = \frac{11.1 \text{ G}}{12.0 \text{ G}} \times 100\% = 92.5\% \text{ W/W}$ The Ring Is
92.5 % W/W Silver. Very Low Concentrations In Studies

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Solution, Mechanical Mixture B) Aqueous Solution,
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