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Study Of Reaction Rates: Clock Reaction Lab

Chemical Reaction- Called A Clock Reaction- Will Be Used To Determine Quantitatively The Influence Of Concentration On Rate. The First Reaction Used Is The Oxidation Of Iodide Ions By Hydrogen Peroxide In Aqueous Solutions: $2\text{I}^{-}(\text{aq}) + \text{H}_2\text{O}_2(\text{aq}) + 2\text{H}^{+}(\text{aq}) \rightarrow \text{I}_2(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$, 2024

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