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React To Form Calcium Phosphate And Water.  $3 \text{Ca(OH)}_2 + 2 \text{H}_3\text{PO}_4 \rightarrow \text{Ca}_3(\text{PO}_4)_2 + 6\text{H}_2\text{O}$  Double Replacement  
7. Copper And Sulfuric Acid React To Form Copper (II) Sulfate And Water And Sulfur Dioxide  $\text{Cu} + 2\text{H}_2\text{SO}_4 \rightarrow \text{CuSO}_4 + 2\text{H}_2\text{O} + \text{SO}_2$  Single Replacement And Acid Base  
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