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14 SECTION 2 Chemical Formulas And EquationsChemical Formulas And Equations

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Period F t i s f e r t f: Form WS5.13 . 1A Percent Yield CHEMICAL FORMULAS AND

EQUATIONS Page 2 Calculate The Percent Yield In Each Of The Problems Below.

(NOTE: Equations May Not Be Balanced.) 1.  $3.25 \times 10^3$  Kg Of Iron III Oxide Are

Treated In A Blast Furnace With  $1.50 \times 10^3$  Kg Of Carbon Monoxide To Form  $1.30 \times$

$10^3$  kg Of Pure Iron.  $[\text{Fe}_2\text{O}_3 + \text{CO} \rightarrow \text{Fe} + \text{CO}]$  ... 3th, 2024.

Calculations With Chemical Formulas And Equations With Chemical Formulas And

Equations 2 | 1 . 2 ... Percentage Composition Example: A Sample Of A Liquid With A

Mass Of 8.657 G Was ... •Smallest Is  $3.722 \times 10^{-3}$  Mole •C = •H = Step 3: Divide

All Number Of Moles By The Smallest 3th, 2024 Chapter 3 Calculations With

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