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They Are Provided To Assist Your Review Of The Topics Covered Chp 11 Of The

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33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$ Actual yield = 1.45g Theoretical yield = 5.95g
 $\text{KClO}_4 \times 1 \text{mol} \times 122.55 \text{g} \times 2 \text{O}_2 \times 32.00 \text{g} \times 1 \text{mol} = 3.11 \text{g O}_2$
% Yield = $\frac{1.45 \text{g}}{3.11 \text{g}} \times 100\% = 46.6\%$
34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$

Percent purity = 62.8%? g pure FeCO_3 Mar 2th, 2024
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