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Stoichiometry Worksheet #2 (mole-mass, Mass-mole Problems) 1. N 2 + 2O 2 → N 2 O 4 A. If 15.0g Of N 2 O 4 Was Produced, How Many Moles Of O 2 Were Required? 15.0g N 2 O 4 1 Mol N 2 O 4 2 Mol O 2 92.0g N 2 O 4 1 Mol N 2 O 4 = 0.326 Mol O 2 B. If 4.0x10-3 Moles Of Oxygen Reacted, How Many Grams Of N 2 Were Needed? 4.0x10-3 Mol O 2 1 Mol N 2 28 ... 2th, 2024

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In A Camping Stove? 2. How Many Moles Of O 2 Molecules Should Be Supplied To Burn 1 Mol Of CH 4 ... Was At One Time Considered For Use As A Rocket Fuel. ... How Many Formula Units Of Calcium Oxide (CaO) 2th, 2024

Stoichiometry Mixed Problems 1011

Mixed Stoichiometry Problems 1. How Many Moles Of H 2 Would Be Required To Completely React With O 2 To Produce 5 Moles Of Water? 5 Mol H 2 2. H2SO 4 + NaOH → Na 2SO 4 + H 2O A. Balance This Equation B. What Mass Of H 2SO 4 Would Be Required To React With 0.75 Mol Of NaOH? 37g 3. What Mass Of NO 2 Is Formed When NO Reacts With 384 G Of O 2? 4th, 2024

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Mass To Mass Stoichiometry Problems Worksheet Answers

SO3 As A Conversion Factor, We Determine The Mass That This Number Of Moles Of SO3 Has. The First Step Resembles The Exercises We Did In Section 6.4 "Mole-Mole Relationships In Chemical Reactions". As Usual, We Start With The Quantity We Were Given: 3.59 Mol Fe2O3×3 Mol SO31 Mol Fe 2O3=10.77 Mol SO3 The Mol Fe2O3 Units Cancel, Leaving Mol SO3 ... 1th, 2024

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