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We Start With The Quantity We Were Given:  $3.59 \text{ Mol Fe}_2\text{O}_3 \times 3 \text{ Mol SO}_3 / 1 \text{ Mol Fe}_2\text{O}_3 = 10.77 \text{ Mol SO}_3$  The Mol  $\text{Fe}_2\text{O}_3$  Units Cancel, Leaving Mol  $\text{SO}_3$  ... Jul 1th, 2024

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