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Carbon Combusts To Form CO₂ And 0.50 G Of The Carbon Combusts To Form CO. The Resultant Mixture Of CO₂ And CO Is Passed Through Excess Na₂CO₃, 2024.

Answer On Question #58182 Chemistry Organic Chemistry ...2) Aniline And Acetanilide Stand In A One-to-one Molar Relationship, So 0.0322 Mol Of Aniline Produces 0.0322 Mol Of Acetanilide. The Amount Of Acetic Anhydride Does Not Play A Role, Since It Is In Excess. Then, $n(\text{Aniline}) = n(\text{Acetanilide}) = 0.0322$. We Find The Molar Mass Of Acetanilide: $M(\text{Acetanilide}) = 12 + 8 + 1 + 9 + 14 + 1 + 16 + 1 + 135 = 187$. 89 M Aceta ... 2th, 2024
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