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Section 3 Classifying Chemical Reactions Decomposition Decomposition Reaction- One Substance Breaks Down, Or Decomposes, Into Two Or More Substances.  $AB \rightarrow A + B$  Most Decomposition Reactions Require The Use Of Heat, Light, Or Electricity. Mar 6th, 2024 Classifying Chemical Reactions Classifying Chemical Reactions Types Of Chemical Reactions: 1. Synthesis 2. Decomposition 3. Combustion 4. Single-Replacement 5. Double-replacement \*Classifying How Atoms Rearrange During A Chemical Reaction. December 12, 2014 1. Synthesis Reaction: 2 Or More Substances React To Produce A Single Product  $A + B \rightarrow AB$  Jun 15th, 2024 Classifying Chemical Reactions Worksheet Classifying Chemical Reactions Worksheet Name: \_\_\_\_\_ Period: \_\_\_\_\_ Classify Each Reaction As Acid/base, Redox, Synthesis, Decomposition, Single Replacement, Double Replacement Or Combustion. They May Be More Than One. The Equations Are Not Balanced. 1. Jan 14th, 2024.

Section 8-3: Classifying Chemical Reactions Section 8-3: Classifying Chemical Reactions Reaction Type General Format Characteristics Of Reactants Characteristics Of Products Example (s) Decomposition  $AB \rightarrow A + B$  Only One Reactant Two Or More Simpler Products  $CuCO_3 (s) \rightarrow CuO + CO_2 (g)$   $2 H_2 O (l) \rightarrow 2 H_2 (g) + O_2 (g)$  Synthesis AKA: Combination  $O (l) + A + B \rightarrow AB$  2 Two Elements Or Two Simple Feb 15th, 2024

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