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### **Colligative Properties- Page 1 Lecture 4: Colligative ...**

B Raises Boiling Point Of Water From 100 °C To 100.64° C •  $\Delta T_f$  Lowers Freezing Point Of Water From 0° C To -2.32° C •  $\pi$  Raises Osmotic Pressure From 1 atm To 28.6 atm While The Temperature Was Changing By A Degree 4th, 2024

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Gizmo Warm-up Adding Salt Or Other Substances To Water Can Affect The Temperature At Which It Freezes Or Boils. These Effects And Others, Known As Colligative Properties, Are Explored In The Colligative Properties Gizmo™. Check That No Solute Is Selected And The Air Temp. Is 25 °C. 1. Describe The Motion Of The Water Molecules In The Beaker. 1th, 2024

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### 12.3 Colligative Properties - REMONDINI

Colligative Properties For Molecular Solute Vs. Ionic Solute. Example: 1.0 M Of Sugar & 1.0 M Of Salt Show Different  $\Delta T$  B And  $\Delta T$  F Even Though The Molality Concentrations Of The Solute Are Identical. Note That;  $\Delta T = M \cdot K$  F,b Where K Is Only A Function Of The Solvent !!! Reason For The Difference In Colligative Properties Must Be Based On ... 3th, 2024

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